

SET A

# INDIAN SCHOOL MUSCAT <br> SECOND TERM - EXAMINATION <br> PHYSICS (042) 

| MARKING SCHEME |  |  |  |
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| $\begin{aligned} & \text { SET } \\ & \text { A } \end{aligned}$ | $\begin{array}{\|l} \hline \text { QN. } \\ \text { NO } \end{array}$ | VALUE POINTS | MARKS SPLIT UP |
|  |  | SECTION A | $3 \times 2=6$ |
|  | 1 | Any four differences between Isothermal and adiabatic process. | $4 \times 1 / 2=2$ |
|  | 2 | (a) On a rainy day, water vapour content in air is more than any other day. Hence on a rainy day, the density of air is lesser. Speed of sound in air is inversely proportional to the square root of its density. Sound travels faster in lighter air than in the heavier one. <br> (b) The distance between a node and the immediate next anti-node is one fourth of a wavelength | 1 1 |
|  | 3. | (a) Wien's displacement law states that the black-body radiation curve for different temperatures will peak at different wavelengths that are Inversely proportional to the temperature. <br> (b) Latent heat of fusion of a solid is defined as the amount of heat required to convert a unit mass of the substance from the solid state to the liquid state without changing the temperature. <br> (OR) <br> (a) Stefan's law of radiation: The quantity of radiant energy emitted by a perfect blackbody per unit time per unit surface area of the body is directly proportional to the fourth power of its absolute temperature. <br> (b)Latent heat of vaporization is defined as the amount of heat required to convert a unit mass of the substance from the liquid state to the vapors state without changing the temperature. | 1 1 1 1 1 |
|  |  | SECTIN - B | $8 \times 3=24$ |
|  | 4. | A motion be Simple harmonic motion only when, <br> 1. Acceleration of particle is just opposite to motion of body <br> 2. Acceleration is directly proportional to displacement e.g., $a=-\omega^{2} x$ |  |



|  |  | (ii) Diatomic gas molecule is made of two atoms. i.e., $N=2$, So $K=1$, <br> Degrees of freedom, $\mathrm{f}=3 \mathrm{~N}-\mathrm{K}$, $\mathrm{f}=3 \times 2-1=5$ | 1 |
| :---: | :---: | :---: | :---: |
|  | 6. | $\begin{aligned} & P V=\frac{2 R T}{R T} \\ & V=\frac{m}{P} \end{aligned}$ <br> For one mole of a gas at STP we have $\begin{aligned} & V=\frac{1 \times 8.314 \times 273}{1.013 \times 10^{5}} \\ & V=0.0224 \mathrm{~m}^{3} 10^{2} .4 \text { litres } \\ & V=22 . \end{aligned}$ | 1 <br> 1 <br> 1 |
|  | 7. | (a) According to first law of thermodynamics: - The change in the internal energy of a closed system is equal to the amount of heat supplied to the system, minus the amount of work done by the system on its surroundings. $\Delta \mathrm{Q}=\Delta \mathrm{U}+\Delta \mathrm{W}$ <br> Where: <br> $\Delta \mathrm{Q}$ is the heat supplied to the system by the surroundings <br> $\Delta \mathrm{W}$ is the work done by the system by the surroundings $\Delta \mathrm{U}$ is the change in internal energy of the system. <br> (b) Internal energy of thermodynamic system is sum of KE and PE of particles. | 1 <br> 1 <br> 1 |
|  | 8. | Statement of Bernoulli's theorem. Diagram <br> Proof of Bernoulli's theorem. | $\begin{gathered} \hline 1 \\ 1 / 2 \\ 11 / 2 \\ \hline \end{gathered}$ |
|  | 9. | From pascal's law $\begin{gathered} \mathrm{P}_{1}=\mathrm{P}_{2} \\ \frac{\mathrm{~F}_{1}}{\mathrm{~A}_{1}}=\frac{\mathrm{F}_{2}}{\mathrm{~A}_{2}} \\ \frac{\mathrm{~F}_{1}}{\pi \mathrm{r}_{1}{ }^{2}}=\frac{\mathrm{F}_{2}}{\pi \mathrm{r}_{2}{ }^{2}} \\ \mathrm{~F}_{1}=\frac{\mathrm{F}_{2} \mathrm{r}_{1}{ }^{2}}{\mathrm{r}_{2}{ }^{2}} \\ \mathrm{~F}_{1}=\frac{1350 \times 9.8 \times\left(5 \times 10^{-2}\right)^{2}}{\left(15 \times 10^{-2}\right)^{2}} \\ \mathrm{~F}_{1}=1470 \mathrm{~N} \\ \mathrm{~F}_{1}=1.47 \times 10^{3} \mathrm{~N} \\ \mathrm{P}_{1}=\mathrm{F}_{1} / \mathrm{A}_{1} \\ \mathrm{P}_{1}=1.9 \times 10^{5} \mathrm{~Pa} \end{gathered}$ <br> (OR) $\mathrm{H}=2 \mathrm{~T} \text { Coso } / \mathrm{rdg}$ <br> Sbstitution of values $=2.8 \times 10^{-2} \mathrm{~N} / \mathrm{m}$ | $1 / 2$ <br> $1 / 2$ <br> $1 / 2$ <br> $1 / 2$ <br> $1 / 2$ <br> 1 <br> 1 <br> 1 |


|  | 10. | Consider a capillary tube of radius $r$ dipped in a liquid of surface tension $\sigma$ and density $\rho$. As the pressure is greater on the concave side of a liquid surface, $\mathrm{p}=\frac{2 \sigma}{R}$ <br> where $R=$ radius of curvature of the <br> concave meniscus <br> Due to this excess pressure, the liquid rises <br> In the capillary tube to height $\mathrm{h}, \mathrm{p} \frac{r}{R}=\cos ; \mathrm{R}=\frac{r}{\cos \theta}$; $\begin{aligned} & \mathrm{p}=\frac{2 \sigma \cos \theta}{r}=\mathrm{h} \rho \mathrm{~g} \\ & \mathrm{~h} \rho \mathrm{~g}=\frac{2 \sigma \cos \theta}{r} \quad ; \quad \mathrm{h}=\frac{2 \sigma \cos \theta}{r \rho g} \end{aligned}$ | $1 / 2$ <br> $1 / 2$ <br> $1 / 2$ <br> 1 <br> $1 / 2$ Dig. |  |
| :---: | :---: | :---: | :---: | :---: |
|  | 11 | (a) Modulus of elasticity is defined as ratio of the stress to the corresponding strain produced, within the elastic limit. $\mathrm{N} / \mathrm{m}^{2}$ <br> (b) Definition of Young's modulus and Bulk modulus of elasticity. <br> (c) Due to long and repeated use of bridge it lost elastic nature Or <br> Due to elastic fatigue. | $\begin{aligned} & 1 / 2+1 / 2 \\ & 1 / 2+1 / 2 \end{aligned}$ $1$ |  |
|  |  | SECTION C | $1 \times 5=5$ |  |
| 1 | 12 | (i) C <br> (ii) B <br> (iii) Any option <br> (iv) D <br> (v) D |  |  |
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